



Synthesis and Antimicrobial Studies of Some New Thiadiazepine Derivatives

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ABSTRACT

A new series of thiadiazepine derivatives 4a-k were synthesized from aldehyde and triazoles. They were characterized by advanced techniques like IR, ¹HNMR and mass spectroscopy and their antimicrobial studies were carried out.

Key words: Thiadiazepine, antimicrobial, spectral analysis, 1, 2, 4-triazoles and heterocyclization.

INTRODUCTION

Currently, there is wide range and spectrum of effective antimicrobials is essential for the future. 1, 2, 4-triazole, a five member heterocyclic with three nitrogen atoms in the ring which are by far the best known class of triazoles, Comprises wide variety of medicinal activities like antifungal, antimicrobial, anticonvulsant¹⁻⁸. 1,2,4-triazole is an important emerging moiety in pharmaceutical study and a lot of work can be carried out on this molecule for obtaining better therapeutic activity. The synthesis of compounds belonging to thiadiazepine series constitute an

important area due to their interesting diverse biological activities such as antibacterial, antifungal, analgesic, anti HIV and antidepressant properties⁹⁻¹⁶

MATERIALS AND METHOD

3-substituted-4-amino-5-mercapto-1,2,4-triazoles 2a-b were synthesis through multistep sequential reaction from different phenols¹⁷. The reaction between triazoles 1 a-b and 1-phenyl-chloro pyrazole gave the twelve new thiadiazepine 4a-k (Scheme 1)

Characterization Data

Compd. No.	R	Mol. Formula Mol. Wt.	Yield (%) m.p. ^o C	Colour and crystal nature	% Analysis Found (Calculated)		
					C	H	N
4a	H	C ₁₃ H ₁₁ N ₆ ClS 318	72 140-43	Yellow crystals	49.07 (49.05)	3.49 (3.45)	26.46 (26.41)
4b	CH ₃	C ₁₄ H ₁₃ N ₆ ClS 332	68 208-10	White crystals	50.64 (50.60)	3.96 (3.91)	25.38 (25.30)
4c	C ₂ H ₅	C ₁₅ H ₁₅ N ₆ ClS 346	85 140 - 45	Yellow crystals	43.38 (43.35)	4.39 (4.33)	24.21 (24.27)
4d	C ₃ H ₇	C ₁₆ H ₁₇ N ₆ ClS 360	76 179-81	White crystals	53.35 (53.33)	4.77 (4.72)	23.36 (23.33)
4e	C ₆ H ₅	C ₁₉ H ₁₅ N ₆ ClS 394	65 247-48	Pale Yellow crystals	57.81 (57.86)	3.86 (3.80)	21.34 (21.31)
4f	4-ClC ₆ H ₄	C ₁₉ H ₁₄ N ₆ Cl ₂ S 429	71 212-14	Yellow crystals	53.22 (53.27)	3.24 (3.27)	19.66 (19.62)
4g	4-CH ₃ C ₆ H ₄	C ₂₀ H ₁₇ N ₆ ClS 418	88 134-36	Yellow crystals	58.88 (58.82)	4.12 (4.16)	20.54 (20.58)
4h	4-OCH ₃ C ₆ H ₄	C ₂₀ H ₁₇ N ₆ OClS 424	73 132-35	Pale Yellow crystals	56.68 (56.60)	4.06 (4.01)	33.09 (33.01)
4i	4-BrC ₆ H ₄	C ₁₉ H ₁₄ N ₆ ClSBr 473	64 110-12	Brown crystals	48.36 (48.30)	2.98 (2.96)	17.71 (17.79)

4j	4-NO ₂ C ₆ H ₄	C ₁₉ H ₁₄ N ₇ O ₂ ClS 439	66 225- 28	Dark Yellow crystals	51.98 (51.93)	3.12 (3.18)	22.36 (22.32)
4k	2-OHC ₆ H ₄	C ₁₉ H ₁₅ N ₆ OSCl 420	74 156-58	White crystals	55.59 (55.60)	3.66 (3.65)	20.47 (20.48)

Solvent for crystallization :Ethanol

Antimicrobial studies (Antibacterial and antifungal activities)

The antibacterial activity of the newly synthesized compounds were carried out against four different pathogenic organisms, two each of Gram-negative and Gram-positive, they are

- i) *Staphylococcus aureus* : (Gram-positive)
- ii) *Bacillus subtilis* : (Gram-positive)
- iii) *Escherichia coli* : (Gram-negative)
- iv) *Pseudomonas aeruginosa* : (Gram-negative)

The fungus is

i) *Candida albicans*

The antibacterial activity and antifungal activity of the newly synthesized compounds in the present investigation has been assessed by minimum inhibitory concentration (MIC) by serial dilution method.

The results of the antimicrobial studies are shown in Table 4.2.

Results

Among the compounds tested for antibacterial and antifungal activity compounds 4b and 4h showed very good activity, which may be due to the presence of chlorine atom in the molecule. Most of other compounds also showed significant activity comparable with that of the standard and in many cases, the activity was much higher.

Table 4.2: Antibacterial and antifungal activity data of compounds 4a-j

Compd. No.	Antibacterial activity (MIC in µg/ mL)				Antifungal activity (MIC in µg/ mL)
	<i>S. aureus</i>	<i>E. coli</i>	<i>P. aeruginosa</i>	<i>B. subtilis</i>	<i>C. albicans</i>
4a	0.25	0.25	0.25	0.25	0.25
4b	0.125	0.125	0.125	0.125	0.125
4c	0.25	0.25	0.25	0.25	0.25
4d	0.25	0.25	0.25	0.25	0.25
4e	0.25	0.25	0.25	0.25	0.25
4f	0.25	0.25	0.25	0.25	0.25
4g	0.25	0.25	0.25	0.25	0.25
4h	0.125	0.125	0.125	0.125	0.125
4i	0.25	0.25	0.25	0.25	0.25
4j	0.25	0.25	0.25	0.25	0.25
Standard Furacin	0.5	0.5	0.5	0.5	-
Standard Flucanazol	-	-	-	-	0.25

RESULTS AND DISCUSSION

δ 0.93(t, 3H, methyl proton of propyl group), 1.67(m, 2H, methylene protons of propyl group), 2.48(s, 3H, methyl protons of pyrazole group), 2.68(t, 2H, methylene protons of propyl group), 7.51-7.63 (m, 5H, phenyl protons), 10.14 (s, 1H, -N=CH proton).

Further the structure has been confirmed by recording the mass spectra of some selected compounds.

The mass spectra of compound 4b and 4d showed (M⁺+1) peak at m/z 333/335 and 361/363 respectively consistent with the molecular formula C₁₄H₁₃N₆ClS and C₁₆H₁₇N₆ClS.

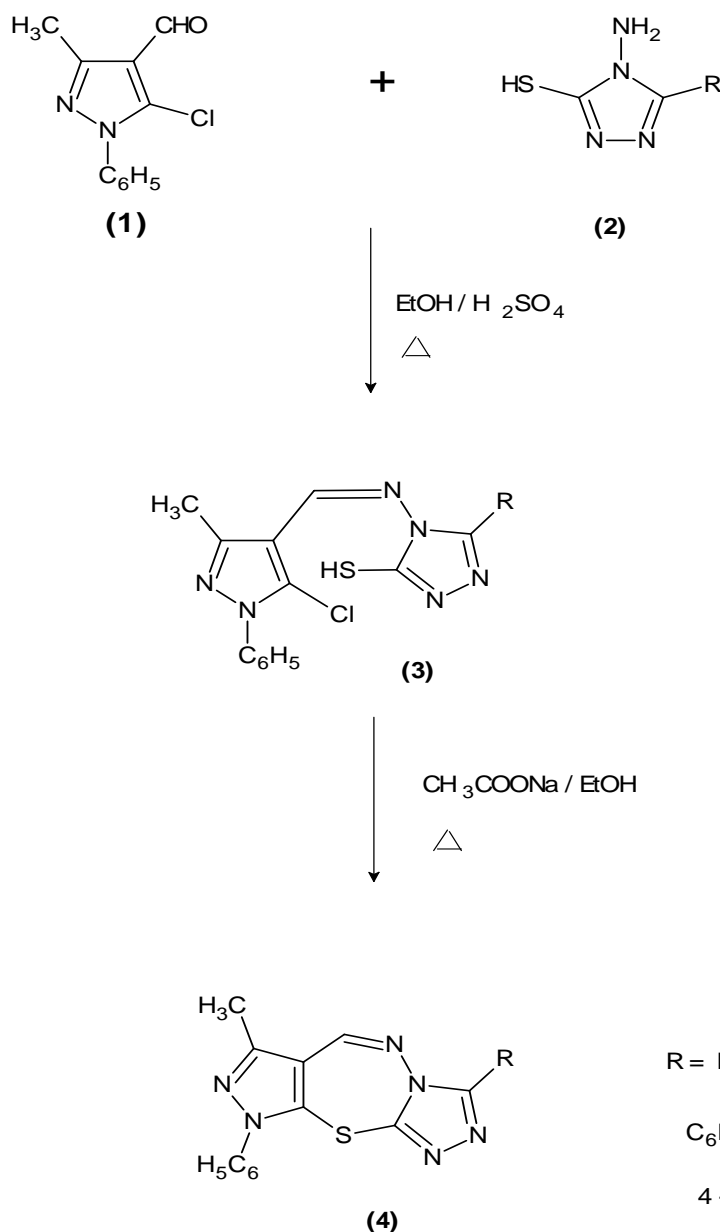
Similarly the structural elucidation of novel triazolothiadiazepines was performed by elemental analysis, IR, ¹H-NMR and mass spectral IR spectra were recorded in a Shimadzu FTIR 8400s spectrophotometer using KBr pellets. H¹ - NMR spectra were recorded in deuterated dimethyl sulphoxide in an AV500 NMR spectrometer with

tetramethylsilane as internal standard. The mass spectra were recorded in a Shimadzu GC MS-Q P5050 mass spectrometer. The elemental analysis of all the compounds done in Flash thermo1112 series CHN analyser gave the values within the permissible limit of 0.4%. Melting point were determined by open capillary and are uncorrected. The IR, $^1\text{H-NMR}$ and MS were consistent with the assigned structures. IR spectra of condensed thiadiazepines displayed disappearance of band at 1665cm^{-1} due to $\text{C}=\text{O}$ str. And 2750cm^{-1} due to SH str. of 5-substituted -4-amino -3-mercapto-1,2,4-triazoles. $^1\text{H-NMR}$ spectra of thiadiazepines showed the C-H proton at 5ppm and-CH proton at 7.5 ppm. The proton attached to nitrogen of thiadiazepine ring

was observed at 8.5 ppm. The solvent peak due to DMSO-d_6 was observed at 2.4ppm.

MATERIALS AND METHOD

1- Phenyl -3-methyl-chloro-pyrazole- 4 - aldehyde (1) was obtained by Vilsmeier-Haack formylation of 1-phenyl-3-methyl- pyrazole -4-one. Condensation of 3-substituted-4-amino-5-mercapto-1,2,4- triazoles (2) with 1-phenyl-3-methyl-5-chloro-pyrazole-4- aldehyde (1) in ethanol medium in presence of conc. sulphuric acid as catalyst gave Schiff's base (3). Schiff's base (3) when treated with anhydrous sodium acetate in ethanol medium gave novel triazolothiadiazepines(4).



R = H, CH₃, C₂H₅, C₃H₇

C₆H₅, 4-CH₃C₆H₄, 4-OCH₃C₆H₄, 4-ClC₆H₄

4-BrC₆H₄, 2-HOC₆H₄, 4-NO₂C₆H₄

CONCLUSIONS

In the present study, our attention was focused on the synthesis and antimicrobial evaluation of a series of thiadiazepines. Based on the resulting biological evaluation data, all the compounds showed mild to moderate antimicrobial potential. Among all the synthesized thiadiazepines, compound 4b and 4h showed broad antimicrobial activity against all tested microbes.

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